

KYNOCH FERTILIZER

SAFETY DATA SHEET

Potassium Nitrate WS

Date Issued / Revised Date : 25 September 2022

New version : 3.0

Date previously revised : 1 February 2021

Replaced version : 2.0

Prepared according to: United Nations GHS (Rev 9E) (2021) and SANS 10234:2019

(This Safety Data Sheet conforms to the requirements set by the Department of Agriculture, Land reform and Rural development of the Republic of South Africa on the 29 March 2022)

SECTION 1: IDENTIFICATION

1.1 GHS product identification

Product Name : Potassium Nitrate WS

¹ GHS - Globally Harmonized System of Classification and Labelling of Chemicals

1.2 Other means of identification

Description : Potassium Nitrate, KNO3, Nitrate of Potassium (NOP)

Chemical name : Potassium Nitrate

CAS Number ² : **7757-79-1** EC Number ³ : **231-818-8**

1.3 Recommended use of materials and restrictions on use

Recommended use of material : Intended to be used as a fertilizer and in fertilizer blends

Description : Source of plant nutrients

Restrictions on use : None Identified

1.4 Supplier's details

Supplier's details : 1st Floor, ETG House

62 Weirda Road East

Sandton

2196

Tel no: (011) 317-2000

1.5 Emergency phone number

Emergency phone number : Dial Triple Zero (000) and ask for fire

: Ambulance or the Fire department – 10177

: Spilltech - 086 100 0366

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² "CAS Number" - CAS Number is a numerical designation for chemicals that is maintained by the Chemical Abstracts Service (CAS) of the American Chemical Society.

^{3 &}quot;EC Number" - The European Community number (EC number) is a unique identifier that was assigned to substances for regulatory purposes within the European Union by the European Commission.

SECTION 2: HAZARD IDENTIFICATION

2.1 Classification of substance or mixture

Product Defined : Substance

Summarized Classification

Types of Hazards	Hazard Class	Category/subcategory	H-Statement
Physical Hazards	Oxidizing Solid	Category 3	H272 ²
Health Hazards	Not Classified 1		
Environmental Hazards	Not Classified 1		

Classification according to the United Nations GHS (Rev 9E) (2021) and SANS 10234:2019

Classification by Organization

Organization	Substance	CAS Number	Classification
EPA-NZ	Potassium Nitrate	7757-79-1	Not Listed
ECHA	Potassium Nitrate	7757-79-1	Oxid. Solid 3
ILO (WHO)	Potassium Nitrate	7757-79-1	No Classification
AICIS	Potassium Nitrate	7757-79-1	No Classification

Reference: (European Chemical Agency [ECHA], n.d.) & (Environmental protection agency [EPA]. New Zealand Government, n.d.) & (The Australian Industrial Chemicals Introduction Scheme [AICIS], n.d.) & (International Labour organization [ILO], n.d.)

2.2 GHS Label elements, including precautionary statements

Pictogram



Pictogram Name : Flame over circle

Signal Word : Danger

Hazard Statements : H272 - May intensify fire; oxidiser.

Precautionary Statements : P210 - Keep away from heat, hot surfaces, sparks, open flames and

other ignition sources. No smoking.

: P220 - Keep away from clothing or other combustible materials.

P280 - Wear protective gloves/protective clothing/eye protection/face

protection/hearing protection/...

Reference: (Pubchem, GHS, n.d.)

2.3 Other hazards that do not result in classification

Hazards : May cause eye, skin and respiratory irritation.

Reference: (European Chemical Agency [ECHA], n.d.) & (Pubchem, search, n.d.)

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¹ "Not Classified" – Data conclusive but not at sufficient levels for classification.

² "H-Statement" – Hazard Statement. Full decryption in Section16

SECTION 3: COMPOSITION / INFORMATION ON INGREDIENTS

3.1 Substance

Common name : Potassium Nitrate
EC Name : Potassium Nitrate
Chemical Formula : KNO₃, HNO3.K
Molecular Weight : 101.103 g/mol

Nutrient Content : 13% Total Nitrogen (N), 13% Nitric Nitrogen (NO₃), 38% Potassium (K)

 CAS Number
 : 7757-79-1

 EC Number
 : 231-818-8

Impurities and stabilizers : N/A

Reference: (European Chemical Agency [ECHA], n.d.) & (The Australian Industrial Chemicals Introduction Scheme [AICIS], n.d.)

3.2 Mixture

Mixture : Not Applicable

SECTION 4: FIRST AID MEASURES

4.1 Description of first aid measures

General information : In case of persisting adverse effects consult a physician.

After inhalation : Supply fresh air. Consult doctor in case of complaints.

After skin contact : Remove affected clothing. Immediately rinse with water (can use mild

soap). If skin irritation continues, consult a doctor.

After eye contact : Rinse opened eye for several minutes under running water (remove contact

lenses if easily possible). Seek medical treatment.

After swallowing : Rinse out mouth. Make victim drink water (maximum of 2 drinking glasses).

Do NOT induce vomiting. If symptoms persist consult doctor.

4.2 Most important symptoms and effects, both acute and delayed

Effects : The substance is irritating to the eyes, skin and respiratory tract. Ingestion

could cause effects on the blood. This may result in the formation of methaemoglobin. The effects may be delayed. Medical observation is

indicated.

Symptoms : Inhalation - Cough. Sore throat.

: Ingestion - Abdominal pain. Blue lips, fingernails and skin.

Dizziness. Laboured breathing. Confusion. Convulsions.

Diarrhoea. Headache. Nausea. Unconsciousness.

: Skin contact - Redness.

: Eye contact - Redness, pain.

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^{1 &}quot;N/A" – Not available

4.3 Indication of any immediate medical attention and special treatment needed

No further relevant information available

SECTION 5: FIRE-FIGHTING MEASURES

5.1 Suitable extinguishing medium

Suitable extinguishing agents

: Water - Use flooding quantities of water for extinction.

Inappropriate extinguishing media

: Do NOT use chemical extinguisher or foam or attempt to smother the fire with steam or sand.

Sand

Foam

Carbon dioxide (CO2)

Dry chemical

Notes : Use fire extinguishing methods suitable to surrounding conditions.

5.2 Specific hazards arise from chemical

Warning : In case of fire, there is a potential option of explosion, especially if

fertilizers are contaminated by inappropriate (incompatible) chemical

substances (e.g., oils, see section 10). Toxic fumes may be formed in fire.

Hazardous Combustion Products : Nitrogen oxides, metal oxide/oxides, ammonia

Fire hazard : Non-flammable substance

Explosion hazard : Not applicable

Reactivity : None

5.3 Special protective action for Fire-Fighters

Special protective actions for firefighters : Promptly isolate the scene by removing all persons from the vicinity of the incident if there is a fire.

: No action shall be taken involving any personal risk or without suitable

training.

Special protective equipment for fire-fighters

: Fire-fighters should wear appropriate protective equipment and selfcontained breathing apparatus (SCBA) with a full face-piece operated in positive pressure mode.

: Clothing for fire-fighters (including helmets, protective boots, and gloves) conforming to European standard EN 469 will provide a basic level of protection for chemical incidents.

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SECTION 6: ACCIDENTAL RELEASE MEASURES

6.1 Personal precautions, protective equipment, and emergency procedures

Percussions : No action shall be taken involving any personal risk or without suitable

training.

Equipment : Wear appropriate respirator when ventilation is inadequate. Put on

appropriate personal protective equipment.

Procedure : Evacuate surrounding areas. Keep unnecessary and unprotected personnel

from entering. Do not touch or walk through spilt material. Provide

adequate ventilation.

6.2 Environmental precautions

Environmental : Avoid dispersal of spilt material and runoff and contact with soil,

waterways, drains and sewers.

: Inform the relevant authorities if the product has caused environmental

pollution (sewers, waterways, soil, or air).

: Discharge into the environment must be avoided.

6.3 Methods and material for containment and cleaning up

Small Spill : Move containers from spill area. Vacuum or sweep up material and place in

a designated, labelled waste container. Dispose of via a licensed waste

disposal contractor.

Large Spill : Move containers from spill area. Approach the release from upwind.

Prevent entry into sewers, water courses, basements, or confined areas. Vacuum or sweep up material and place in a designated, labelled waste

container. Dispose of via a licensed waste disposal contractor.

6.4 Reference to other sections

Section 7 : Information on safe handling.

Section 8 : Information on personal protection equipment.

Section 13 : For disposal information.

SECTION 7: HANDLING AND STORAGE

7.1 Precautions for safe handling

Handling

: Ensure adequate ventilation. Avoid ingestion and inhalation. Avoid dust formation. Wear protective gloves/eye protection/face protection/. Do not get in eyes, on skin, or on clothing. Wash hands thoroughly after handling.

: Further processing of solid materials may result in the formation of combustible dusts. The potential for combustible dust formation should be taken into consideration before additional processing occurs. Provide appropriate exhaust ventilation at places where dust is formed.

: For precautions see section 2.2.

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¹ PPE – Personal precautions, protective equipment.

7.2 Conditions for safe storage, including any incompatibilities

Storerooms and receptacles

: No special requirements.

One common storage facility

: Avoid contamination, particularly with incompatible substances: flammable materials and lubricants, oxidising agents, acids, bases, sulphides, chlorates, chlorines, chromates, nitrates, permanganates; Metal powders, e.g., copper, nickel, cobalt, zinc, and their alloys).

: Incompatible products: Separate from reducing agents and combustible materials. Keep away from acids or bases. On farm keep away from hay, grain, diesel, etc.

: Incompatible materials : Sources of ignition. Direct sunlight.

Handling of product

: Minimize dust generation.

Keep container tightly closed.

Room conditions

: Keep in a dry, well-ventilated place. Recommended storage temperature at < 30°C. (Room temperature). DO NOT expose the substance to temperatures above 50 °C.

: Protect against humidity (product is hygroscopic and tends to cake or disintegrate)

Storage Class

: (TRGS 510): 10 - 13 Other liquids and solids: Non-Combustible Solids

Reference: (BAUA, 2016)

7.3 Specific end use(s)

Specific end use(s)

: Apart from the uses mentioned in section 1.3 no other specific uses are stipulated

SECTION 8: EXPOSURE CONTROL AND PERSONNEL PROTECTION

8.1 Control Parameters							
	Compound	Cas Number		TWA 1	STEL ²		
South African Labour Department	Potassium Nitrate	7757-79-1		Not Listed	Not Listed		
International Labour organization (ILO)	Potassium Nitrate	7757-79-1		Not Listed	Not Listed		
OCHA	Potassium Nitrate	7757-79-1		Not Listed	Not Listed		

¹ TWA – Long term exposure: Time Weighted Average (8-hour period)

Reference: (South African Labour Department, 2021) & (ILO, n.d.) & (OSHA, n.d.)

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² STEL – Short term exposure: Short term exposure limit (15 min period)

Routes of exposure

: he substance can be absorbed into the body by inhalation of its aerosol and by ingestion.

Inhalation risk

: Evaporation at 20°C is negligible; a harmful concentration of airborne particles can, however, be reached quickly when dispersed.

Effects of short-term exposure

: The substance is irritating to the eyes, skin and respiratory tract. Ingestion could cause effects on the blood. This may result in the formation of methaemoglobin. The effects may be delayed. Medical observation is indicated.

Effects of long-term or repeated

exposure

: Not Listed

Reference: (ILO, n.d.)

8.2 Appropriate engineering controls

: Ensure adequate ventilation, especially in confined areas. Ensure that eyewash stations, and safety showers are close to the workstation location. See Section7.

8.2 Individual protection measures

Eye/face protection

: Wear safety glasses.

Use equipment for eye protection tested and approved under appropriate government standards. SABS adoption: SANS 50166:2018(SA), EN 166(EU) or NIOSH (US).

Skin Protection

: Handle with gloves.

Gloves must be inspected prior to use. Use proper glove removal technique (without touching glove's outer surface) to avoid skin contact with this product. Dispose of contaminated gloves after use in accordance with applicable laws and good laboratory practices. Wash and dry hands.

Body Protection

: Choose body protection in relation to its type, to the concentration and amount of dangerous substances, and to the specific work-place. The type of protective equipment must be selected according to the concentration and amount of the

dangerous substance at the specific workplace.

Respiratory protection

: Not required under normal conditions of use.

Where protection from nuisance levels of dusts is desired, use type N95 (US) or type P1 (EN 143) dust masks. Use respirators and components tested and approved under appropriate government standards such as NIOSH (US) or CEN

(EU).

Control of environmental exposure

No special environmental precautions required



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SECTION 9: PHYSICAL AND CHEMICAL PROPERTIES

9.1 Properties

Physical state : Solid 1

: Substance² Composition

Colour **Colourless or White**

Odour : Odourless

: 335°C Melting point/freezing point Boiling point or initial boiling point

and boiling range

: 400°C.

Flammability : Not flammable

Lower and upper explosion

limit/flammability limit

Not determined

Flash point : The study does not need to be conducted because the flash point is only

relevant to liquids and low melting point solids.

: Not a self-heating substance Auto-ignition temperature

Oxidizing Properties : Yes (Oxidizing solid, Category 3), Based on the transport classification,

Class 5.1. O2. PG III, it was concluded that potassium nitrate crystals have oxidising properties. Prilled potassium nitrate do not have oxidizing properties according to the outcome of the conical pile UN Test O.1 result.

Decomposition temperature : 400°C

рΗ : ca. 7 @ 20 °C

Kinematic viscosity : The study does not need to be conducted because the substance is a solid.

100 g/I water @ 25°C Solubility

Partition coefficient: n-octanol/water

(log value)

: The study does not need to be conducted because the substance is

inorganic.

: The study does not need to be conducted because the melting point is Vapour pressure

above 300°C.

: 2.1 q/cm³ @ 20°C Density and/or relative density

Relative vapour density : Not determined

Bulk Density (Volumetric) : Not Listed

MMAD= 222.1 μ m; 0.21%, 10%, 50% and 90% is < 10.00 μ m, < 94.6 μ m, < Particle characteristics

211.8 μ m and < 424.9 μ m, resp.

Molecular Formula KNO³

101.103 g/mol Molecular Weight

Reference: (ECHA, n.d.) & (Pubchem, search, n.d.)

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¹ "Solid" – Is a substance that cannot be classified as a liquid or Gas.

² "Substance" – Is chemical elements and their compounds in their natural state or obtained by production process)

SECTION 10: STABILITY AND REACTIVITY

Reactivity : Contact with combustible materials may cause fire.

Chemical stability : Stable under normal conditions.

Hazardous Reactions : A dangerous reaction will not occur.

Conditions to Avoid : Avoid contact with incompatible materials. Avoid heat, flame, and sparks.

Incompatible Materials : Combustible materials, reducing agents, acids, and alkalis.

Hazardous Decomposition Products : These products are nitrogen oxides

SECTION 11: TOXICOLOGY

11.1 Acute Toxicity

Classification : No Classification

Description : Based on the available data, magnesium nitrate does not have to be

classified according to the CLP Regulation with regard to acute toxicity.

Method	Compound	Cas Number	Measure	Value	Subject
Oral	Potassium Nitrate	7757-79-1	LD50 ¹	>2000 mg/kg bw ²	Rat
Inhalation	Potassium Nitrate	7757-79-1	LC50	>527 mg/L air	Rat
Dermal	Potassium Nitrate	7757-79-1	LD50	>5000 mg/kg bw	Rat

¹ "LD50" – Lethal Doses. The dosage at which 50% mortality was observed.

Reference: (ECHA, n.d.) & (Pubchem, search, n.d.) & (EPA. New Zealand Government, n.d.)

11.2 Skin corrosion/irritation

Classification : No classification

Description : P0018 (ammonium nitrate) was regarded as non-irritant to rabbit skin.

Subjects : Rabbits

Reference: (ECHA, n.d.)

11.3 Serious eye damage/irritation

Classification : No Classification

Description : A single application of the test item to the non-irrigated eye of two rabbits

produced moderate conjunctival irritation. One treated eye appeared normal at the 72-Hour observation and the other treated eye appeared

normal at the 7-Day observation.

Subjects : Rabbits

Reference: (ECHA, n.d.)

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² "bw" - body-weight/day

11.4 Respiratory or skin sensitisation

Classification : No classification

Description : Since there was no indication that the test substance elicits an SI ≥ 3 when

tested up to 50%, Natriumnitrat HQ unbehandelt (non-food grade) was

considered not to be a skin sensitizer.

Subjects : Mouse

Reference: (ECHA, n.d.)

11.5 Germ cell mutagenicity

Classification : No classification

Description : Potassium nitrate is not mutagenic in Salmonella typhimurium strains TA

1535, TA 1537, A 1538, TA 98, TA 100, and TA 92 with and without metabolic

activation. No chromosomal aberrations were induced in a Chinese hamster fibroblast cell line without metabolic activation.

Subjects : Salmonella typhimurium strains

Reference: (ECHA, n.d.)

11.6 Carcinogenicity

Classification : No Classification

Description : Potassium nitrate is not genotoxic and no substance related neoplastic

lesions were observed in the chronic toxicity study. There is no positive

correlation between nitrate intakes and the incidence of cancer.

Reference: (ECHA, n.d.)

11.7 Reproductive toxicity

Classification : No classification

Description : Classification for reproductive toxicity is not required based on a NOAEL of

1500 mg/kg bw in a screening study and supporting evidence showing

absence or low toxicity in the first generation; Classification for

developmental toxicity is not required based on a NOAEL of 1500 mg/kg

bw.

Subjects : Rat

Reference: (ECHA, n.d.)

11.8 STOT ² - single exposure

No data available

² "STOT" - Specific target organ toxicity.

Reference: (ECHA, n.d.) & (Pubchem, search, n.d.)

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11.9 STOT 2- repeated exposure

No Classification

A reliable short-term repeated dose toxicity study with potassium nitrate is available. This OECD 422 study did not show any effects up to the highest dose level tested (1500 mg/kg bw/day) when administered orally.

Based on an expert statement, a sub-chroning repeated dose toxicity study does not need to be conducted.

Reference: (ECHA, n.d.) & (Pubchem, search, n.d.)

11.10 Aspiration hazard

No data available

Reference: (ECHA, n.d.) & (Pubchem, search, n.d.)

11.11 Route of Exposure and potential effects

Swallowing : Abdominal pain. Blue lips, fingernails and skin. Dizziness. Laboured

breathing. Confusion. Convulsions. Diarrhoea. Headache. Nausea.

Unconsciousness.

Inhalation : Cough. Sore throat.

Eye exposure : Redness, pain

Skin exposure : Redness

Reference: (ILO, n.d.)

11.12 Long- and short-term effects

No data available

Reference: (ECHA, n.d.)

SECTION 12: ECOLOGICAL INFORMATION

12.1 Toxicity

Classification : Triggers for classification are not met.

Aquatic Toxicity

Compound	Cas Number	Organism	Species	Time	Measure	Value
Potassium Nitrate	7757-79-1	Fish	Oncorhynchus Mykiss	96-h	LC50 ¹	> 100 mg/L
Potassium Nitrate	7757-79-1	Fish	Fathead minnow	32-d	NOEC ²	58 mg/L
Potassium Nitrate	7757-79-1	Aquatic invertebrates	Daphnia magna	96-h	EC50 ¹	490 mg/L
Potassium Nitrate	7757-79-1	Aquatic invertebrates	Daphnia magna	12-d	NOEC	>245 mg/L
Potassium Nitrate	7757-79-1	Aquatic Algae and Cyanobacteria	benthic diatoms	10-d	EC50	>1700 mg/L
Potassium Nitrate	7757-79-1	Micro-organisms	Activated sludge of a predominantly domestic sewage	3-h	EC50 ¹	>1000 mg/L

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² "STOT" - Specific target organ toxicity.

Terrestrial Toxicity

Potassium Nitrate

Compound	Cas Number	Organism	Species	Time	Measure	Value
Potassium Nitrate	7757-79-1	Macro-organisms				Not Tested
Potassium Nitrate	7757-79-1	Arthropods				Not Tested
Potassium Nitrate	7757-79-1	Plant				Not Tested
Potassium Nitrate	7757-79-1	Micro organisms				Not Tested

¹ "LC50 /EC50" - (Median Lethal Concentration/Median Effective Concentration) They are the concentrations at which 50% mortality or inhibition of a function (e.g., growth or growth rate) was observed.

Birds

Reference: (ECHA, n.d.) & (Pubchem, search, n.d.)

7757-79-1

12.2 Persistence and degradability

Stability

: The substance does not hydrolyze nor is there evidence for photodegradation.

Potassium nitrate is an inorganic salt, soluble in water. It is a neutral salt; the ions have little tendency to react with water. Hydrolysis is therefore not a relevant parameter for this substance and testing does not appear scientifically necessary.

Not Justified

Biodegradation

Readily biodegradation study does not need to be conducted since the substance is inorganic.

Reference: (ECHA, n.d.)

12.3 Bioaccumulate potential

Description

: Simple inorganic salts with high aqueous solubility will exist in a dissociated form in an aqueous solution. Such a substance has a low potential for bioaccumulation.

Reference: (ECHA, n.d.)

12.4 Mobility in soil

Adsorption

: The study does not need to be conducted because the physicochemical properties of the substance indicate that it can be expected to have a low potential for adsorption

Nitrate is not bound to the soil and will follow water movements. Nitrate can therefore leach when the soil receives more water than it can take up. This happens (in) mainly in the late autumn, winter, and early spring. There exist a lot of studies on the environmental impact of NO₃ and NH₄+/NH₃.

: Volatilization is unlikely due to the properties of the substance.

Reference: (ECHA, n.d.)

Volatilization

12.5 Other adverse effects

Classification : No data available

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² "NOEC" - No Observed Effect Concentration. NOEC is the highest tested concentration for which there are no statistically significant difference of effect when compared to the control group.

³ "ECx" - It is the concentrations at which x % (10% for EC10) effect was observed or derived statistically when compared to the control group.

SECTION 13: DISPOSAL CONSIDERATIONS

The generation of waste should be avoided or minimized wherever possible. Disposal of this product, solutions and any by-products should at all times comply with the requirements of environmental protection and waste disposal legislation and any regional local authority requirements. Dispose of surplus and non-recyclable products via a licensed waste disposal contractor. Waste should not be disposed of untreated to the sewer unless fully compliant with the requirements of all authorities with jurisdiction. Waste packaging should be recycled. Incineration or landfill should only be considered when recycling is not feasible. This material and its container must be disposed of in a safe way. Empty containers or liners may retain some product residues. Avoid dispersal of spilled material and runoff and contact with soil, waterways, drains and sewers.

SECTION 14: TRANSPORT INFORMATION

12.1 UN Modelled regulations

UN Number : 1486

UN proper shipping name : Potassium Nitrate

Transport hazard class(es) : 5.1 – Oxidizer

Label :



Packing group : III - Substances presenting low danger

Environmentally hazardous : No classification

Special precautions: : ADR/RID - 34, B120, IB8, IP3, T1, TP33

IMDG² - 208, 967 IATA³ - A83

Transport in Bulk according to IMO : Not specified

instructions

Reference: (Hazmat Tool. n.d.) & (BAM. 2021)

SECTION 15: REGULATORY INFORMATION

15.1 Safety, Health, and environmental regulations specific for the substance or mixture

Regulations : This Safety Data Sheet conforms to the requirements set by the

Department of Agriculture, Land reform and Rural development of the Republic of South Africa, United Nations GHS (Rev 9E) (2021) and SANS

10234:2019, on the 29 March 2022.

Restrictions : The substance is not subjected to any prohibitions or restriction in South

Africa.

Chemical Safety Assessment: : For this product a chemical safety assessment was not carried out.

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¹ ADR/RID - International Carriage of Dangerous Goods by Rail (RID) and by Road (ADR)

² IMDG - The International Maritime Dangerous Goods (IMDG)

³ IATA - International Air Transport Association (IATA)

SECTION 16: OTHER INFORMATION

16.1 Preparation and revision

Latest Version

Version Number Ver. 3

Preparation Date 25 August 2022

Where the changes as made Complete overall of all data to comply with GHS regulations

Previous Version

Version Number Ver. 2

Preparation date February 2021

16.2 Abbreviations and Acronyms

GHS Globally Harmonized System of Classification and Labelling of Chemicals

European Chemical agency **ECHA**

The Australian Industrial Chemicals Introduction Scheme AICIS

EPA-NZ Environmental protection agency New Zealand

ILO (WHO) International labour organization (World health organization)

CAS Number CAS Number is a numerical designation for chemicals that is maintained by the Chemical Abstracts

Service (CAS) of the American Chemical Society.

EC Number The European Community number (EC number) is a unique identifier that was assigned to

substances for regulatory purposes within the European Union by the European Commission.

H-Statement Hazard Statement

P-Statement Precautionary Statements

Causes serious eye irritation Hazard Statements H319

Wash hands [and ...] thoroughly after handling. Precautionary Statements P264

P265 Do not touch eyes.

P280 Wear protective gloves/protective clothing/eye protection/face protection/hearing

protection/...

Not Applicable N/A

Not Classified Data conclusive but not at sufficient levels for classification

PPE Personal precautions, protective equipment.

Time Weighted Average TWA Occupational Exposure Limits OFI STOT Specific target organ toxicity

(Median Lethal Concentration/Median Effective Concentration): They are the concentrations at which LC50 / EC50

50% mortality or inhibition of a function (e.g., growth or growth rate) was observed.

NOEC (No Observed Effect Concentration) NOEC is the highest tested concentration for which there are no

statistically significant difference of effect when compared to the control group

It is the concentrations at which x % (10% for EC10) effect was observed or derived statistically when

compared to the control group LD0 Lethal Dose 0, represents the dose at which no individuals are expected to die.

LC0 Lethal concentration 0, represents the concentration at which no individuals are expected to die. I DI o

Lethal dose low, is the lowest dosage of a compound that is introduced to the human body or that of

an animal by any means apart from inhalation that will cause the death of the individual.

16.3 References

ECx

BAM. (2021)Retrieved **Dangerous** Goods Database. From https://www.dgg.bam.de/quickinfo/en/show/g3dgamgyon62sdvxmoz2nhhc3m

(The BAM offers with the expert portal TES information and service concerning the transport and packaging of dangerous substances and goods as well as explosives act)

BAUA. (2016). Technical Rule for Hazardous Substances. TRGS 510 Storage of hazardous substances in non-stationary containers. Retrieved from https://www.baua.de/EN/Service/Legislative-texts-and-technical-rules/Rules/TRGS/TRGS-510.html

(The German Federal Institute for Occupational Safety and Health offers selected publications in English, Baua's research aims to ensure a safe and healthy working environment that is adapted to the needs of humans.)

Date Issued: 12-7-2022, Version 3.0 Previously Issued: 1-2-2021, Version 2.0 Page 14 | 16 **Environmental protection agency [EPA]. New Zealand Government. (n.d.)** Database search. Chemical Classification and Information Database (CCID). Retrieved from https://www.epa.govt.nz/search/SearchForm?Search=

(EPA-Environmental protection agency. EPA is the government agency responsible for regulating activities that affect Aotearoa New Zealand's environment.)

European Chemicals Agency [ECHA]. (n.d.) Information on Chemicals. Retrieved from https://echa.europa.eu/registration-dossier/-/registered-dossier

(ECHA - European Chemicals Agency. The European Chemicals Agency, is an agency of the EU. They implement the EU's chemicals legislation to protect your health and the environment. There work also contributes to a well-functioning internal market, innovation, and the competitiveness of Europe's chemicals industry.)

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(ILO-International Labour organization. ILO is a specialized agency of the United Nations. The database data was prepared by an international group of experts on behalf of ILO and WHO, with the financial assistance of the European Commission. © ILO and WHO 2021.)

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(The Australian Industrial Chemicals Introduction Scheme (AICIS) helps protect Australians and the environment by assessing the risks of industrial chemicals and providing information to promote their safe use. Focus mainly on heath aspects.)

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16.4 Disclaimer

The information contained in this SDS does not constitute a risk assessment, and should not replace the user's own assessment of risks as required by other health and safety legislation.

This SDS summarises at the date of issue our best knowledge of the health, safety and environmental hazard information related to the product and in particular how to safely handle, use, store and transport the product. Since Kynoch cannot anticipate or control the conditions under which the product may be handled, used, stored, or transported, each user must, prior to usage, review this SDS in the context of how the user intends to handle, use, store or transport the product and beyond, and communicate such information to all relevant parties.

We shall not assume any liability for the accuracy or completeness of the information contained herein or any advice given unless there has been gross negligence on our part.

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